

## Potential use of cyanophyta species *Oscillatoria limnetica* in bioremediation of organophosphorus herbicide glyphosate

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### Abstract:

One locally microalga species *Oscillatoria limnetica* was isolated from the artificial canal around University of Babylon in Al-Hilla city. The isolated microalga species was cultivated in BG-11 media for biomass production and to test their ability to degrade glyphosate that is the widely used organophosphorus herbicide by high performance liquid chromatography analysis. HPLC appeared to be an effective technique to follow decrease of glyphosate in biodegradation process.

Removal efficiencies of glyphosate were evaluated and compared. The results confirmed the high efficiencies of *O.limnetica* for the removal of the target contaminant.

There were significant differences in the rate of glyphosate removal from the medium after 3, 5, 7, 14 and 35 days in all treatments. It was 97.5% after seven days at 5 mg/l treatment, 57.9% and 37.1% after fourteen days at 10 mg/l and 15 mg/l treatments, respectively and 99.9% after thirty five days at 20 mg/l treatment. Maximum removal efficiency percentages of isolated species were increased to reach their maximum in all treatments after thirty five days.

**Keywords:** Cyanophyceae, *Oscillatoria limnetica* , Herbicide , Glyphosate Bioremediation.

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### Introduction

As a result of human influence, the amounts of organic compounds found in surface water have increased in the recent decades. Of these organic compounds, pesticides are most commonly detected in all aquatic environments [1]. Although the use of pesticides increases the productivity of crop and control unwanted organisms, but also increase the

contamination of water resources. This pollution can reduce water quality and lead to significant economic effects when pollutant concentrations warrant closure or mandate remediation of a contaminated drinking water source [2 ; 3]. This contamination stems from the action of rainwater and irrigation water that can take these compounds and carry them to rivers and lakes that are used in cities for water supply [4].

Glyphosate-based herbicides are the world's leading broad spectrum, post-emergent and non-selective herbicides for the control of annual and perennial weeds in agricultural lands, ornamental and residential gardens and in aquatic systems[4]. In the last decades glyphosate consumption has increased sharply and it has become one of the most used agrochemicals globally [5]. Glyphosate is a competitive inhibitor of the enzyme 5-enolpyruvylshikimate 3-phosphate synthase EPSPS), which is involved in the pathway of synthesis of shikimic acid [6].

A treatment that promises to be efficient, economic and safe is bioremediation. This kind of treatment has been approached from a biotechnological point of view in order to be able to have a methodology that is carried out at lower temperature and pressure and requires less energy than the conventional physical-chemical treatment technology, as well as avoiding additional damages to the environment [7]. Bioremediation may provide a suitable method for treatment of water pollution with pesticides effectively [8].

Microalgae are eukaryotic, as the green algae (Chlorophyta) or prokaryotic photosynthetic microorganisms, as the blue-green algae (cyanophyta) [9]. They are considered to be the principal agents for the degradation of pesticides in bioremediation processes. Their cell size, morphology, density and activities show an important role in the uptake and removal of pesticides. High surface area to biovolume ratio of algae provides greater potential for sorption and subsequent interaction with pesticides [10].

Therefore, this study was conducted to investigate the survival and tolerance of cyanophyta species (*Oscillatoria limnetica*) with different concentrations of glyphosate, as well as evaluating their efficiency for removing and recovering this pesticide from contaminated water by high performance liquid chromatography (HPLC) analysis.

## **Material and Methods**

### **Sample collection and identification**

Samples of freshwater algae were collected from artificial canal around University of Babylon in Al-Hilla city by using phytoplankton net.

Phytoplankton net (mesh size 0.50  $\mu\text{m}$ ) which made of silk cloth [11] was swept on surface water and samples were transferred into sterile plastic bottles/containers which marked with data and location of sampling then carried to the laboratory at once to be incubated in suitable and controlled conditions for algae growth [12]. Experimental cultures were incubated at cool white fluorescent lamps (200)  $\mu\text{E}/\text{m}^2/\text{s}$  with a light/dark cycle of 16/8 h. and  $26 \pm 1^\circ\text{C}$  [13]. Moreover the cultures were mildly shaken by hand on alternate days [14].

### **Preparation and sterilization of media**

Modified BG-11 medium was used for the enhancement of cyanophyta growth [15] quoted from [16]. The stocks were prepared for all macro and micro elements (Table1) and dissolved in one liter of distilled water to prepare BG-11 culture medium. Micronutrient solution was sterilized by filtration through 0.22-mm polycarbonate membrane to avoid interaction and precipitation of heavy metals [17]. BG-11 medium was autoclaved at  $121^\circ\text{C}$  for 15min. when cooled, pH was adjusted to approximately 7.4 with 1N NaOH and 1N HCl using pH meter to avoid heavy precipitation [18].

Table 1. BG-11 medium components

Components	Stock solution g.L <sup>-1</sup> dH <sub>2</sub> O	Quantity Used	Concentration in Final Medium (M)
NaNO <sub>3</sub>		1.5 g	1.76 X 10 <sup>-2</sup>
K <sub>2</sub> HPO <sub>4</sub> · 3H <sub>2</sub> O	40	1 ml	1.75 X 10 <sup>-4</sup>
MgSO <sub>4</sub> · 7H <sub>2</sub> O	75	1 ml	3.04 X 10 <sup>-4</sup>
CaCl <sub>2</sub> · 2H <sub>2</sub> O	36	1 ml	2.45 X 10 <sup>-4</sup>
Na <sub>2</sub> CO <sub>3</sub>	20	1 ml	1.89 X 10 <sup>-4</sup>
Citric acid	6	1 ml	3.12 X 10 <sup>-5</sup>
EDTA (disodium salt)	1	1 ml	2.79 X 10 <sup>-6</sup>
Ferric ammonium citrate	6	1 ml	~3 X 10 <sup>-5</sup>
Trace metals mix A5	As shown below	1 ml	
Trace metals mix A5	Amount g	Concentration in Final Medium (M)	
H <sub>3</sub> BO <sub>3</sub>	2.86	4.63 X 10 <sup>-5</sup>	
MnCl <sub>2</sub> ·4H <sub>2</sub> O	1.81	9.15 X 10 <sup>-6</sup>	
ZnSO <sub>4</sub> ·7H <sub>2</sub> O	0.22	7.65 X 10 <sup>-7</sup>	
NaMnO <sub>4</sub> ·2H <sub>2</sub> O	0.39	1.61 X 10 <sup>-6</sup>	
CuSO <sub>4</sub> ·5H <sub>2</sub> O	0.079	3.16 X 10 <sup>-7</sup>	
Co(NO <sub>3</sub> ) <sub>2</sub> ·6H <sub>2</sub> O	0.049	1.70 X 10 <sup>-7</sup>	
Distilled water	1 L		

## Algae isolation and purification:

### Serial dilution culture (SDC) technique

Serial dilutions of the collected samples were prepared starting with 1ml of sample inoculated into 9 ml of BG-11 media. This procedure was repeated with examining of each dilution until one species of algae were obtained. After the target dilution was microscopically examined several times to confirm obtaining unialgal culture, 2 ml of unialgal culture was transferred into 20 ml of fresh BG-11 enhancement solution then incubated under suitable conditions for algal growth till the culture turn into greenish color [19,20]. Obtained algal isolates were identified with the help of classical algal classification references [21, 22].

### Agar plating technique

BG-11 nutrient medium solidified by 2% agar-agar and autoclaved, after sterilization, BG-11 media with 45-50 C° was poured into petri-dishes and left to solidify. Then the surface of each plate was inoculated with 1 ml of sampled water, the inoculum distributed with a sterile spreader or streaking using a sterile loop. The inoculated plates were kept under suitable and controlled conditions for algae growth in a cooled illuminated incubator with about 200 µE/m<sup>2</sup>/s light intensity and 16:8 light: dark period for two weeks [13]. Aggregated colonies were noticed on the surface of the plates. Part of these colonies was cultured on other plates. Each subculture was examined interally, this method was repeated till an unialgal

culture or cultures have been obtained [23]. A small part of unialgal culture (which was microscopically confirmed as unialgal culture) was transferred into BG-11 nutrient solution within a 250 ml sterile flask and incubated for 2-3 weeks according to the method of Jawad [19] to get appropriate growth.

### **Alga cultivation for biomass.**

Prepared 10 ml of isolated alga was grown in 250ml Erlenmeyer flask containing 100 ml of BG-11 medium for two weeks, then transferred to 1000 ml of medium and incubated for 14 days; finally the growth transported to glass pools 5L for mass culture [24 , 25].

### **Pesticide**

The organophosphorus pesticide used in this study is the formulation of the herbicide glyphosate commercially available as Roundup® (containing 480 g active ingredient/L of glyphosate) was obtained from Al-Farah company, Iraq.

### **Experimental Design**

The selected algal isolates were batch-cultured in 500 ml Erlenmeyer flasks. Into each flask 200 ml of liquid culture media, BG11 medium for *O.limnetica* ,was added. Glyphosate was added to the culture medium to the final concentrations 5 , 10 , 15 , 20 mg/l. The flasks were cultivated under the conditions described above. After 3 , 5 , 7, 14and 35 days, 5 ml of algal cultures was filtrated by Millipore filter 0.45 µm. The algal filtrate was used to determine glyphosate residues in the culture medium by high performance liquid chromatography (HPLC).

### **HPLC analysis of glyphosate.**

Analysis of residual glyphosate was done using HPLC system (shimadzu, japan) with a chromatographic column 80SAX from GMB, Germany (length: 25cm, diameter: 4.6mm, particle size: 5 µl) and UV detector at 195 nm. The isocratic mobile phase consisted of 0.8437 g of KH<sub>2</sub>PO<sub>4</sub> dissolved in 960 ml distal water and 40 ml of MeOH, and the pH was adjusted to 2.1 with 85% orthophosphoric acid at a flow rate of 1 ml/min.

The standards and samples (50 µl) were injected into the HPLC system after they were filtered by Millipore filter paper (0.45 µm. pore size) and the time required by a sample to pass through the system is recorded as its retention time. The concentration of residential glyphosate was determined by the following formula:

$$C \text{ sample} = \frac{C \text{ stander} \cdot A \text{ sample}}{A \text{ stander}} \quad C: \text{concentration, A: area}$$

### **Analysis of glyphosate removal**

In agreement with the methodology described by [26]. The percent reductions of glyphosate were calculated by the equation:

$$\text{Parameter reduction (\%)} = \frac{A - B}{A} * 100$$

A = Initial concentration of glyphosate , B= Concentration after treatment.

### **Statistical Analysis**

The general treatment structure was used as an experimental design. Data were analyzed to study bioremediation capacity of *O. limnetica* and Least significant difference (LSD) was used to compare the significant difference between means at p<0.05.

Data present in Table 2 illustrated that *O. limnetica* have the ability to biodegrade glyphosate at different concentrations and significant differences were recorded between initial and residual concentrations after 3, 5, 7, 14 and 35 days.

Glyphosate was reduced from an initial value of about 5 mg/l to 1.95 mg/l, 0.13 mg/l and 0.12 mg/l which corresponds to 60.89%, 97.33% and 97.55% removal efficiency after three, five and seven days of treatment, respectively. In 10mg/l treatment, glyphosate concentration was reduced to 7.17 mg/l, 6.78mg/l, 6.66mg/l and 4.20, which equating to a removal efficiency of approximately 28.25 %, 32.13 %, 33.90 and 57.9% after three, five, seven, and fourteen days of treatment, respectively. In addition, 15mg/l reduced to 12.56 mg/l, 11.30 mg/l, 10.35 mg/l and 9.42 mg/l, which equates to 16.22%, 24.64 %, 30.98 % and 37.18 % removal efficiency after three, five, seven and fourteen days of treatment, respectively. While 20 mg/l reduced to 9.86 mg/l, 4.63 mg/l and 0.02 mg/l which equates to 50.68 %, 76.82 % and 99.9% removal efficiency after seven, fourteen and thirty five days of treatment, respectively. After thirty five days there was no glyphosate in all treatments because the peak of glyphosate disappeared after that time.

In agreement with these results [27] reported that mixed culture of *Spirulina* spp. showed a remarkable capacity to degrade the widely used organophosphorus herbicide glyphosate, that served as sole source of phosphorus for cyanophytal growth. Also, [28] study the tolerance of six cyanophytal strains to the herbicide up to millimolar concentrations. Four strains (*Arthrospira fusiformis*, *Leptolyngbya boryana*, *Microcystis aeruginosa* and *Spirulina platensis*) were able to use the phosphonate as the sole phosphorus (P) source for growth while two strains (*Anabaena* sp. and *Nostoc punctiforme*) were found to possess a glyphosate-insensitive form of EPSP synthase, which is in line with the findings of [29] who reported that tolerance of *Anabaena variabilis* ATCC 29413 is related to the presence of a resistant form of the target enzyme EPSP. [30] too stated that the existence of phosphorous-dependent glyphosate transformation with marine cyanophyta species *Trichodesmium erythraeum*.

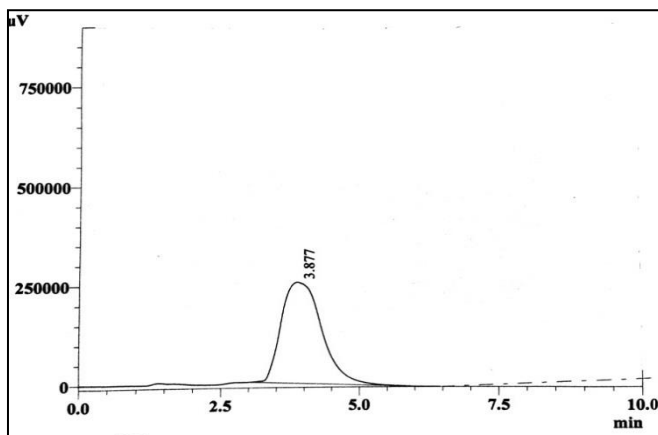
In general, data obtained from Foriani et al. [28] suggested that some hydrolytic enzymes inside the cyanophyta cells may be able to distinguish the herbicide molecule as a substrate, even though with low efficiency. The resulting low rate of C–P bond hydrolysis could not be able to destroy high levels of glyphosate, but may be sufficient to provide the cell with lower quantities of phosphate that are required for growth in the absence of extra physiological phosphorus sources. According to them, the initial tolerance up to 20 mM is in line with the previous reports [31 ; 32] on varying resistance of cyanophyta to this herbicide. On the other hand, the low permeability of the cell membrane to the phosphonate seems to represent another mechanism that allows some cyanophyta strains to grow in the presence of the herbicide. At micromolar concentrations, those at which diffusion across the membrane does not work [33], glyphosate incorporation appears to proceed through a low affinity carrier, and low amounts of herbicide are incorporated into the cell.

**Table 2.** The residual concentration of glyphosate after (3,5,7,14,35) days and *O.limnetica* removal efficiency(%).

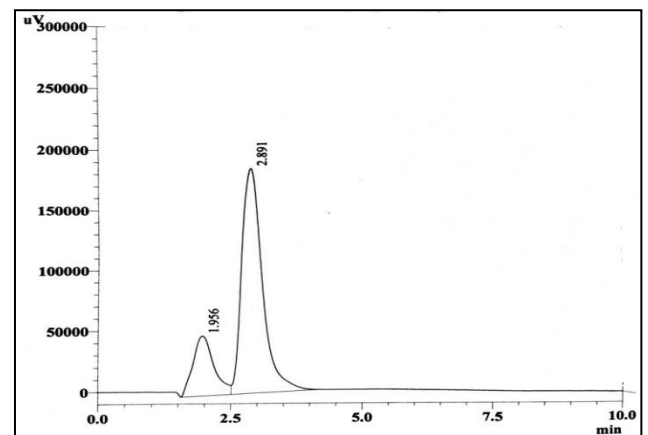
Exposure time (days)	5mg/l		10mg/l		15mg/l		20mg/l	
	Residual Concentration mg/l	Removal Efficiency (%)	Residual Concentration mg/l	Removal Efficiency (%)	Residual Concentration mg/l	Removal Efficiency (%)	Residual Concentration mg/l	Removal Efficiency (%)
3	1.95	60.89	7.17	28.25	12.56	16.22	-----	-----
5	0.13	97.33	6.78	32.13	11.30	24.64	-----	-----
7	0.12	97.55	6.66	33.90	10.35	30.98	9.86	50.68
14	-----	-----	4.20	57.9	9.4	37.1	4.63	76.82
35	-----	-----	-----	-----	-----	-----	0.02	99.9
<b>LSD value</b>	0.941*		0.813*		0.817*		0.941*	

• (p < 0.05)

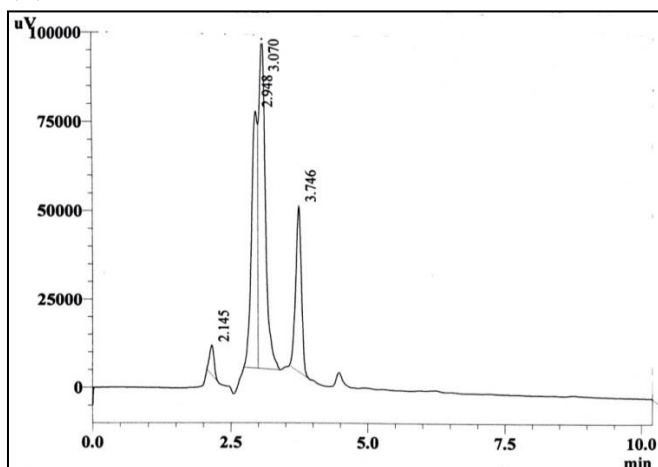
A significant differences between treatments and residual concentration after all days.



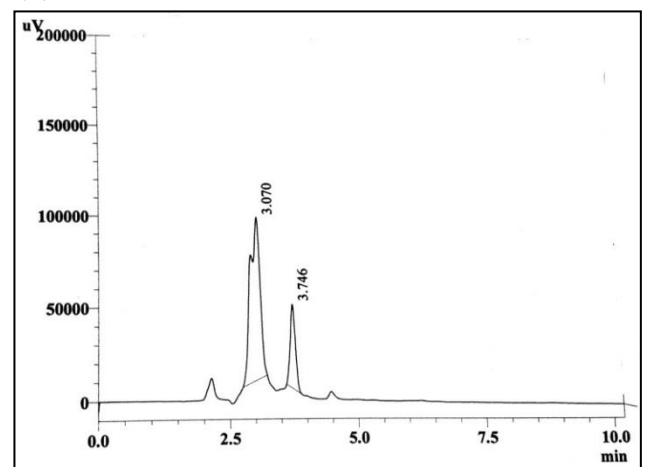
(A)



(B)

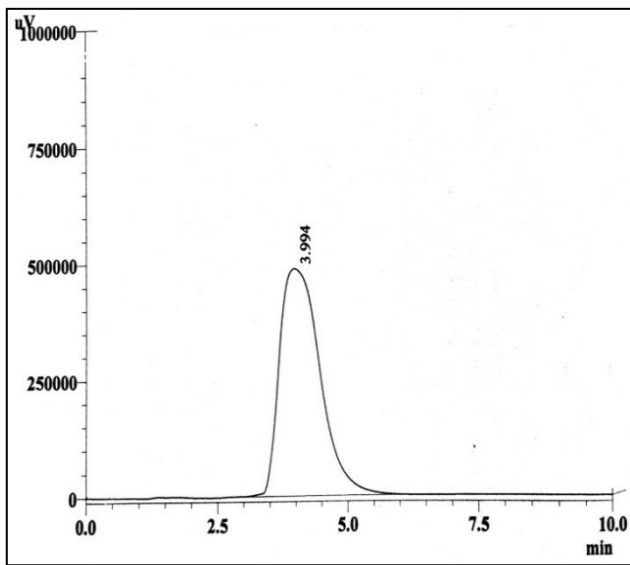


(C)

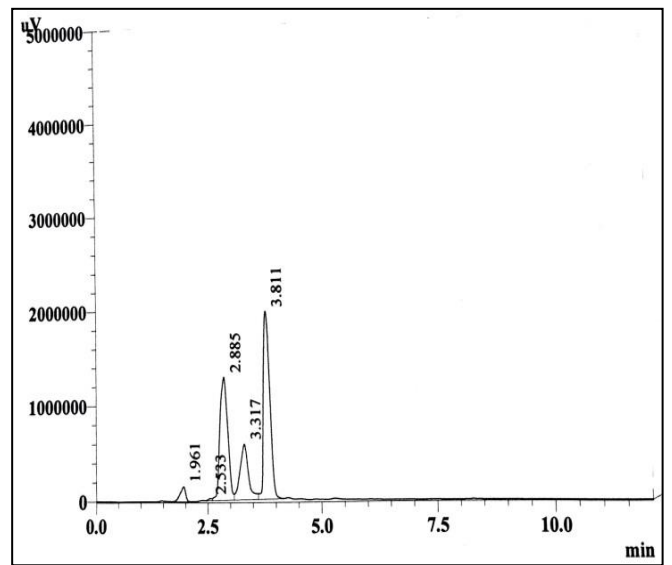


(D)

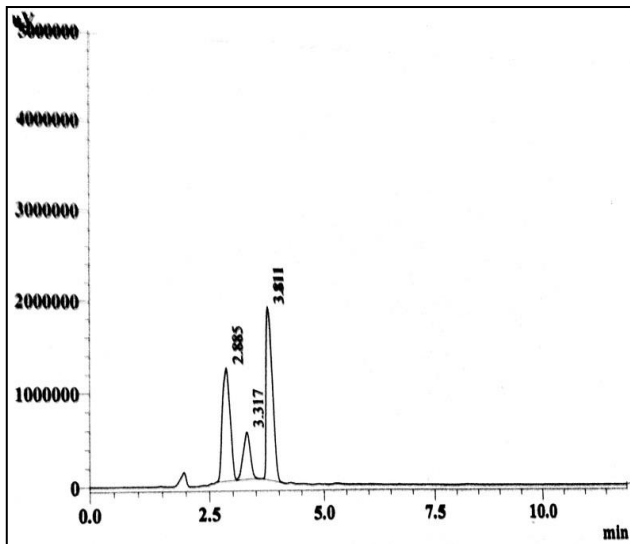
**Fig. 1:** Chromatogram of HPLC for Glyphosate standard solution at 5 mg/l (A) and sample of *O.limnetica* at 5 mg/l glyphosate concentration after 3 days (B), 5 days (C), 7 days (D).



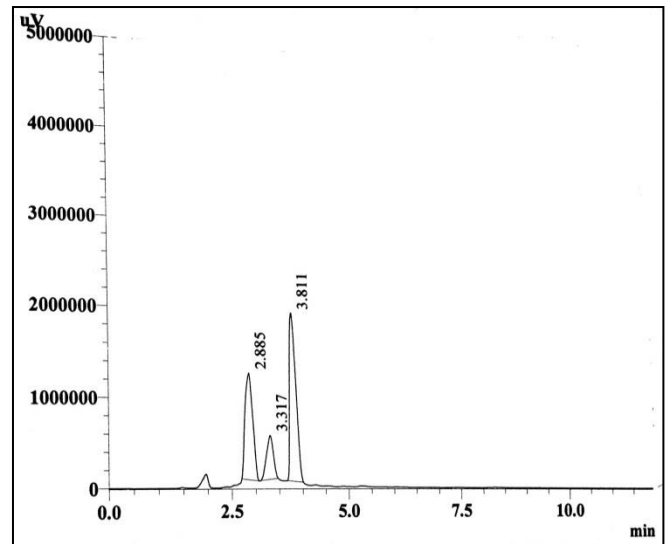
(A)



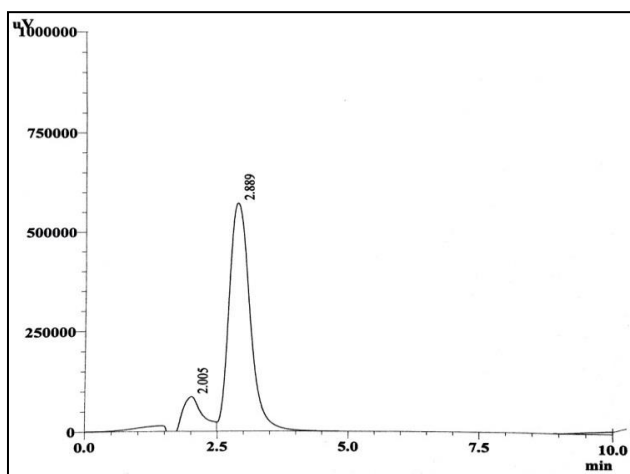
(B)



(C)

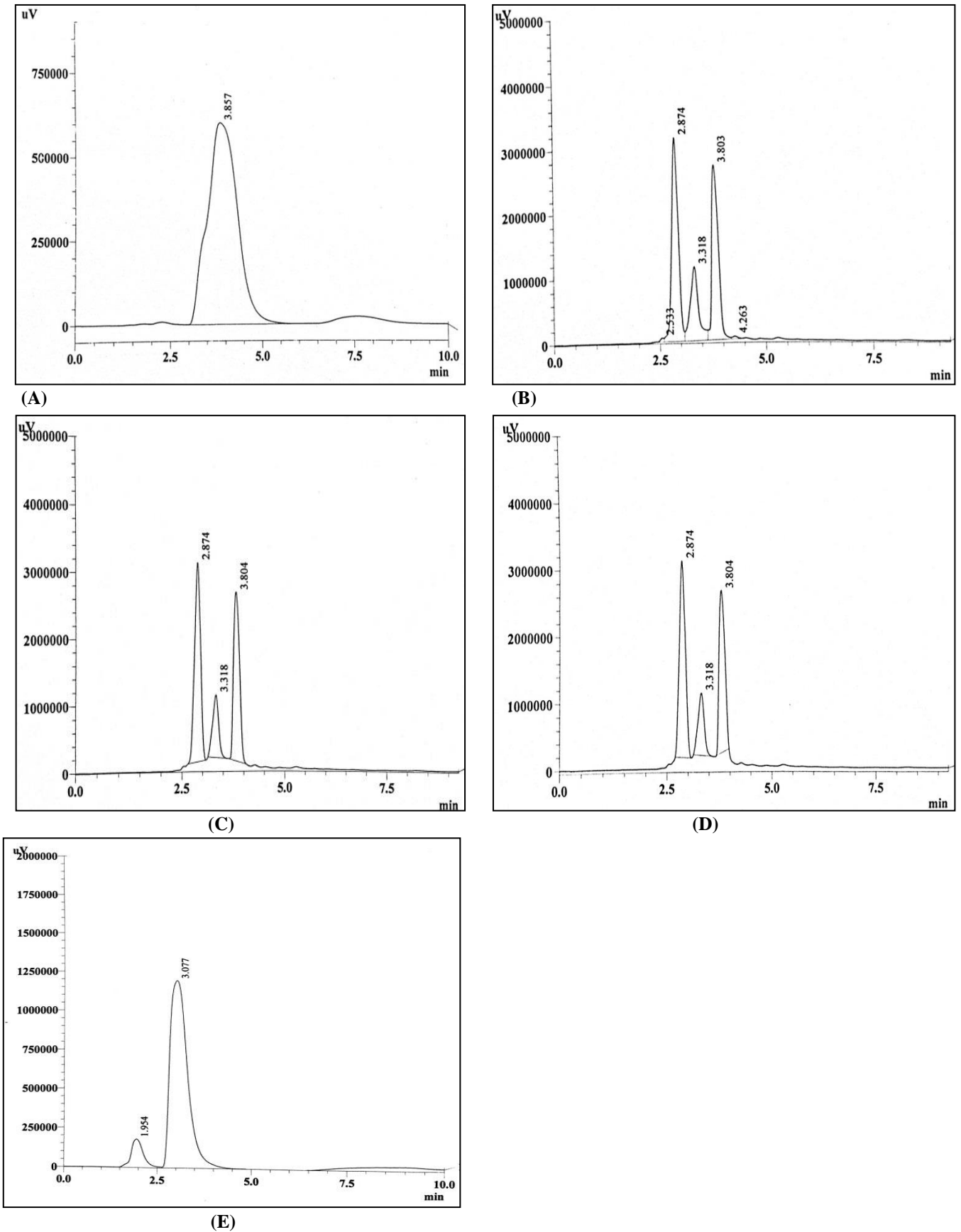


(D)



(E)

Fig. 2: Chromatogram of HPLC for glyphosate standard solution at 10 mg/l (A) and sample of *O. limnetica* at 10 mg/l glyphosate concentration after 3 days (B), 5 days (C), 7 days (D), 14 days (E).



**Fig. 3:** Chromatogram of HPLC for glyphosate standard solution at 15 mg/l(A) and sample of *O.limnetica* at 15 mg/l glyphosate concentration after 3 days (B), 5 days (C), 7 days (D), 14 days (E).



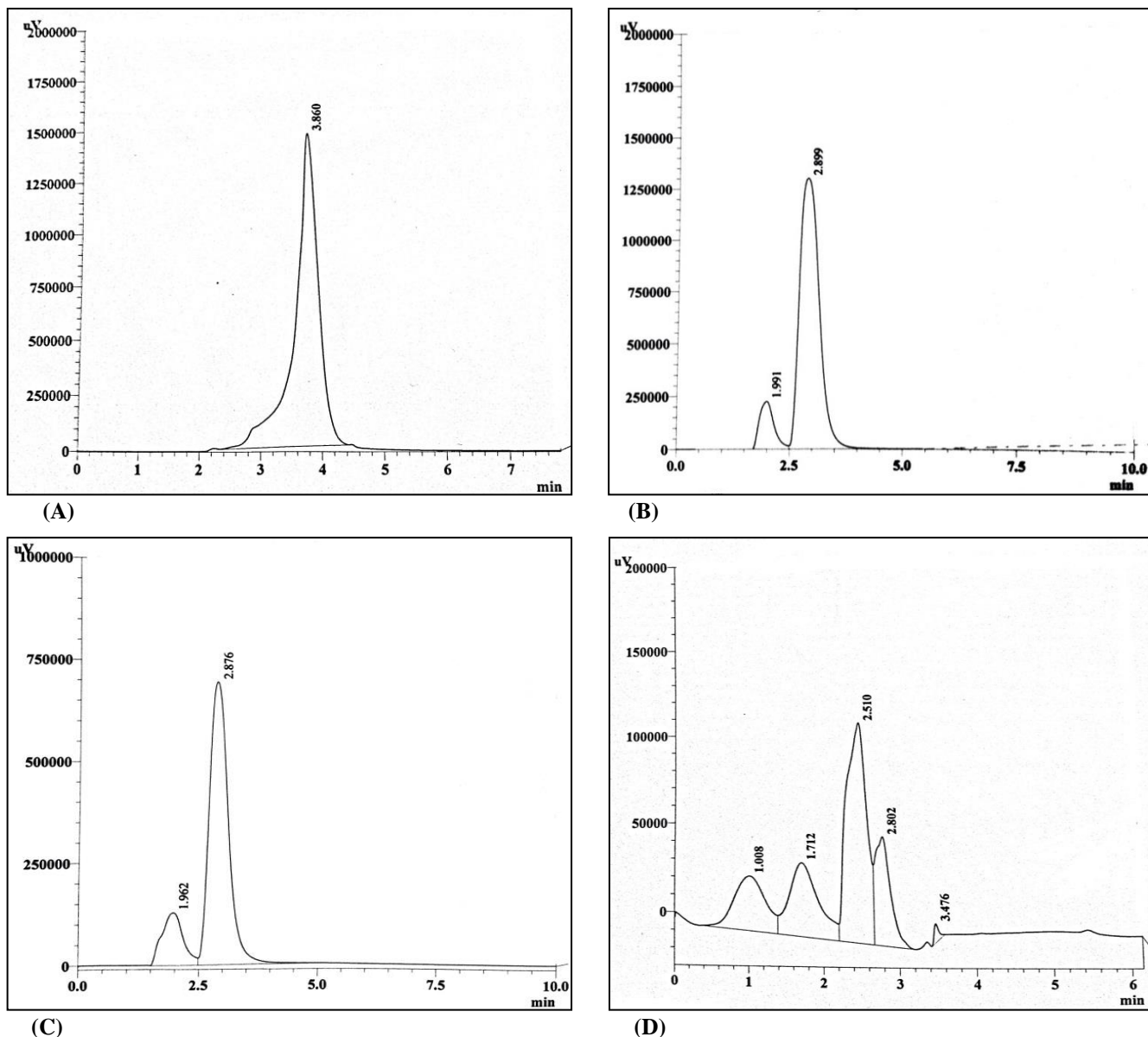


Fig. 4:Chromatogram of HPLC for glyphosate standard solution at 20 mg/l (A) and sample of *O.limnetica* at 20 mg/l glyphosate concentration after 7 days (B), 14 days (C), 35 days (D).

## Conclusions

- (1) The capacity of *Oscillatoria limnetica* to survive and grow in the high concentrations level of the herbicide, show that this strain may possess potential to be used in bioremediation of glyphosate-contaminated environments. In addition, this work adds to the list of glyphosate degrading organisms a new degrading species that is *O. limnetica*
- (2) Glyphosate degradation was measured by HPLC. The method's sensitivity is suitable and reliable to evaluate the presence of glyphosate in water samples.
- (3) Glyphosate removal efficiency percentages of *O.limnetica* were increased to reach their maximum in all treatments after thirty five days.

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